

On the Influence of Moisture on Dielectric Properties of Polyetheretherketone (PEEK) Carbon-Fiber Composites

by Bruce K. Fink, Roy L. McCullough, and John W. Gillespie Jr.

ARL-TR-2236 June 2000

Approved for public release; distribution is unlimited.

The findings in this report are not to be construed as an official Department of the Army position unless so designated by other authorized documents.

Citation of manufacturer's or trade names does not constitute an official endorsement or approval of the use thereof.

Destroy this report when it is no longer needed. Do not return it to the originator.

Abstract

An analysis of local and global mechanisms of heat generation and distribution in carbon-fiber-based composites subjected to an alternating magnetic field has shown that heating is dependent upon the dielectric properties of the polymer matrix. These properties were investigated as functions of temperature, frequency, and moisture content. The results indicate little dependence of the dielectric constant on temperature or frequency, while the loss tangent exhibited a substantial dependence on both frequency and temperature. A substantial dependence of loss tangent on moisture content in polyetheretherketone (PEEK) was found.

Table of Contents

		Page
	List of Figures	v
	List of Tables	vii
1.	Introduction	1
2.	Review of Dielectric Properties	3
3.	Characterization	9
4.	Summary	16
5.	References	19
	Distribution List	21
	Report Documentation Page	41

INTENTIONALLY LEFT BLANK.

List of Figures

<u>Figure</u>		Page
1.	Results of Dried-Laminate Induction-Heating Test	2
2.	General Relationship Between Real and Imaginary Parts of Complex Dielectric Constant in Dielectric Materials	6
3.	Room Temperature Dielectric Constant Results for PEEK	10
4.	Room Temperature Dielectric Loss Tangent Results for PEEK	11
5.	Room Temperature Results for Resistivity of PEEK	11
6.	Percent Weight Loss of PEEK Specimens vs. Drying Time	13
7.	Average Experimental Loss Tangent Data for Undried PEEK Specimens	14
8.	Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 24 hr	14
9.	Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 48 hr	15
10.	Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 72 hr	15
11.	Experimental and ICI-Reported Loss Tangent Data for PEEK at 100 kHz	16

INTENTIONALLY LEFT BLANK.

List of Tables

<u>Table</u>		<u>Page</u>
1.	Averaged Data From Three Electrical Property Measurements on PEEK at Room Temperature	. 10
2.	Weight Change Data for PEEK's Dielectric Loss Tangent vs. Moisture Content Testing	12

INTENTIONALLY LEFT BLANK.

1. Introduction

An investigation of induction heating in carbon-fiber-based composites [1] revealed that the heating resulting from an applied alternating magnetic field is due to dielectric losses in the polymer material that separates crossing fibers in a cross-ply or angle-ply laminate, as long as dielectric breakdown does not occur. The analysis included therein assumed that constant values of the fundamental dielectric properties were known. Heating in a dielectric material subjected to an alternating electric field was shown to be linearly dependent upon these material properties—dielectric constant and dissipation factor (or loss tangent). A global model has been developed [2] that established the electromagnetic response of cross-ply laminated composites to alternating magnetic fields. Three fundamentally separate submodels that consider the in-plane response, the through-thickness response, and the global generation of heat and its quantification as the surface temperature profile were developed. One of the necessary inputs to these models is the complex dielectric constant of the polyetheretherketone (PEEK) polymer. PEEK is the matrix material used in the composite prepreg product APC-2 used in the experimental work and theoretical analysis described by Fink et al. [1].

In a preliminary study, a carbon-fiber-reinforced PEEK cross-ply ([0/90]_S) laminate was characterized at various drying times. Ten thermocouples were used to measure the transient temperature profile on the surface of the flat laminated plate, which was heated by magnetic induction using a 10.2-cm-diameter (4 in) Helmholtz coil centered on a 10.2-cm-square (8 in) specimen. The magnetic field was applied to the specimen in five different states of drying: (1) "as-is" (ambient moisture content), (2) 24 hr, (3) 48 hr, (4) 72 hr, and (5) 96 hr at 120 °C and 28-inHg vacuum. Since the same specimen was used for all tests, the drying times were started over after each previous test. Despite PEEK's low reported moisture content of 0.2 to 0.5% [3–5], the average equilibrium temperature exhibited significant decreases with increased drying time, as shown in Figure 1. This apparent dependence of heat production on small changes in moisture content motivated a closer examination of the effects of moisture content on the properties of the material that affect heating under an applied alternating magnetic field.

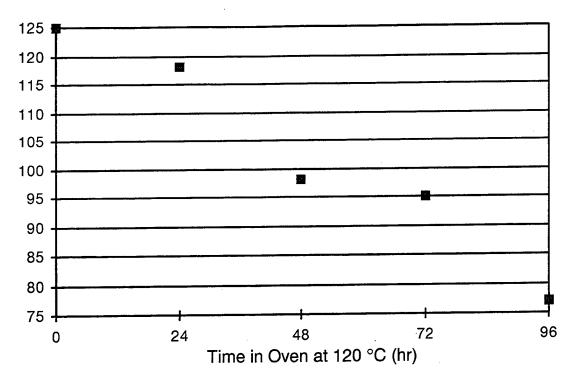


Figure 1. Results of Dried-Laminate Induction-Heating Test.

An equivalent form of an equation for heat generation in our induction heating model [1, 2] can be written as

$$W_{j} = \frac{\beta_{j} V_{j}^{2}}{h}, \tag{1}$$

where W_j is the heat generation at some node j in the plane of the laminate; V_j is the potential difference between the fibers at node j; h is the distance through which the electric field created by V acts; and β_j is a function of several material, environmental and microstructural properties at node j.

$$\beta_i = \omega \varepsilon_0 \kappa \tan \delta \,, \tag{2}$$

where ω is the angular frequency, ε_0 is the permittivity of vacuum, κ is the relative dielectric constant of the polymer, and $\tan\delta$ represents the imaginary part of the complex dielectric

constant of the polymer. Of the components of equations (1) and (2), only the real and imaginary parts of the complex dielectric constant, κ and $\tan\delta$, are potentially functions of moisture content in the composite's matrix material.

In general, the complex dielectric constant can be significantly affected by the frequency, temperature, and moisture content, as well as by the presence of mobile ions in polymeric systems [6]. Values for these properties are typically reported in the literature at room temperature and line frequency without mention of moisture content or the presence of other ionic species. Reported values are not always applicable at the frequencies often used in induction heating of PEEK-based composites (10³ to 10⁶ Hz range) and at temperatures ranging from room temperature to beyond the melting temperature of PEEK (>340 °C).

Compilations of dielectric constants and dissipation factors [7, 8] have been published for various polymers but are difficult to keep current. Manufacturers often supply data for specific materials at specific temperatures and frequencies; however, it is often necessary to characterize these properties for the particular process parameters of interest. This paper discusses the relation of structure to dielectric properties, describes the methods of experimental measurement performed, and reports the results with respect to temperature, frequency, and moisture content for PEEK.

2. Review of Dielectric Properties

Polymeric materials can be divided into two groups: (1) nonpolar and (2) polar. Nonpolar materials exhibit high resistivity, a low dielectric constant, and a negligible loss tangent; they are, for the most part, unaffected by temperature, frequency, and humidity. Polar polymers generally have lower resistivities and higher dielectric constants and loss tangents.

The dielectric constant (unit capacitance, permittivity) and dissipation factor are very different for nonpolar and polar materials as functions of temperature and frequency. The values of both properties for nonpolar materials are nearly constant over a wide frequency range and for

temperatures below the softening point. For polar materials (often called lossy materials), the dielectric constant decreases with an increase in frequency, whereas the dissipation factor increases and decreases cyclically.

Under the influence of an alternating electric field, electric dipoles are oriented in the material such that the magnitude of the induced charges increases as the polarization of the dielectric increases. There are several possible mechanisms of polarization in a dielectric material:

- (1) Electric Polarization—a shift in the center of charge of the electron cloud relative to the positive center of charge on the nucleus when an electric field is applied.
- (2) Ionic Polarization—displacement of positive and negative ions in relation with one another in respect to their equilibrium positions.
- (3) Orientation Polarization—the alignment of permanent dipoles that exist in a material in the absence of an applied field.
- (4) Space Charge Polarization—the alignment of mobile charges in the material whose movements are restricted by interfaces or other barriers in the material.

The total polarization can be expressed as the sum of these four mechanisms; however, the contribution of each source is dependent on frequency, temperature, and applied voltage. Of the four mechanisms of polarization, orientation polarization has the greatest effect on the total dissipation of energy at the frequencies considered in this work (at least for a "clean" polymer that is free of excessive foreign ionic species).

Frequency effects on polarization are related to the time dependency of the polarization mechanisms with respect to their response to an applied field. For example, electronic polarization would be expected to occur rapidly, while permanent dipole orientation would be

slow in comparison. This is usually explained by the difference in the mass of the electrons vs. the molecules and also on the number of conformations and free volume available for motion in polymers. In thermoplastics, orientation polarization relates to the conformational motion of polymeric chains and ionic polarization to the existence of ionic species or other groups either left over from the processing of the polymer or, possibly, deposited by treated fibers during resin impregnation in composite materials.

Molecular movements take a finite time, and complete orientation as induced by an alternating current may or may not be possible, depending on the frequency of the alternating electric field. Thus, at zero frequency, the dielectric constant will be at a maximum and will remain approximately constant until the dipole orientation time is of the same order as the reciprocal of the frequency. Dipole movement will now be limited, and the dipole polarization effect and the dielectric constant will be reduced. As the frequency further increases, the dipole polarization effect will tend to zero and the dielectric constant will tend to be dependent only on the electronic polarization. When there are two dipole species differing in ease of orientation, there will be two points of inflection in the dielectric constant vs. frequency curve. The dielectric constant of unsymmetrical molecules containing dipoles (polar molecules) will be dependent on the internal viscosity of the dielectric.

Besides a dielectric constant dependence on temperature and frequency, polar molecules exhibit relatively high dielectric power losses at certain frequencies; the maximum power losses corresponding to the point of inflection in the dielectric constant vs. frequency curve (Figure 2). At very low frequencies, the dipole movements are able to keep in phase with changes in the electric field and power losses are low. As the frequency is increased, the point is reached when the dipole orientation cannot be completed in the time available and the dipole becomes out of phase. Measures of the fraction of energy absorbed per cycle by the dielectric from the field are the "power factor" and "dissipation factor." The delay between the changes in the field and the change in polarization leads to a current in a capacitor leading the voltage across it when a dielectric is present. The angle of lead is known as the phase angle and given the symbol θ . The value $\theta = \theta$ 0 is known as the loss angle and is given the symbol θ 0. The power factor is defined as

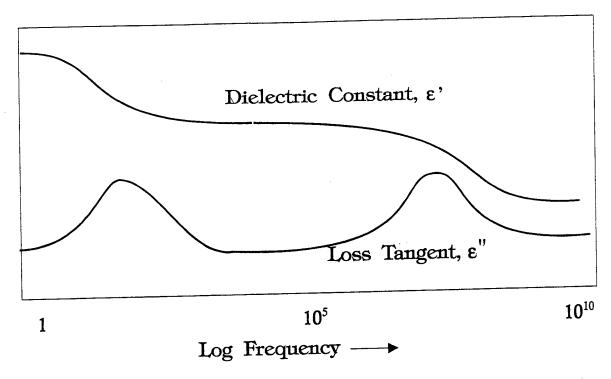


Figure 2. General Relationship Between Real and Imaginary Parts of Complex Dielectric Constant in Dielectric Materials.

 $cos(\theta)$ and the dissipation factor as $tan(\delta)$. When δ is small, the two are equivalent. The "loss factor" is the product of the dissipation factor and the dielectric constant.

At low frequencies, when power losses are low, these values are also low, but they increase when such frequencies are reached that the dipoles cannot keep in phase. After passing through a peak at some characteristic frequency, they fall in value as the frequency further increases. This is because, at such high frequencies, there is no time for substantial dipole movement and the power losses are reduced. Because of the dependence of the dipole movement on the internal viscosity, the power factor, like the dielectric constant, is strongly dependent on temperature.

In polar polymers, the situation is more complex since there are many dipoles attached to one chain. Either these dipoles may be attached to the main chain, or the polar groups may not be directly attached to the main chain and the dipoles may, to some extent, rotate independently.

In the case with dipoles integral with the main chain, in the absence of an electric field, the dipoles will be randomly disposed but will be fixed by the disposition of the main chain atoms. On application of an electric field, complete dipole orientation is not possible because of spatial requirements imposed by the chain structure. Furthermore, in the polymeric system, the different molecules are coiled in different ways and the time for orientation will be dependent on the particular disposition. Thus, whereas simple polar molecules have a sharply defined power loss maxima, the power loss vs. frequency curve of polar polymers is broad, due to the dispersion of orientation times.

When dipoles are directly attached to the chain, their movement will obviously depend on the ability of chain segments to move. Thus, the dipole polarization effect will be less significant below the glass transition temperature than above it. PEEK is such a polymer; the offset oxygen on the ketone segment provides the polarizability of the polymer. It is expected that the loss tangent will have a significant increase directly after the glass transition temperature, which is about 143 °C.

Ionic polarizability is the inducement of dipoles by relative measurement of positive and negative ions in a partial ionic solid. Ionic polarization generally has a predominate influence on the loss tangent at higher frequencies than for dipole polarization. Since the imaginary part of the complex dielectric constant (or dielectric loss factor) is large whenever the frequency reaches the upper limit for any dielectric response mechanism, such as ionic polarization, a second peak in the loss tangent vs. frequency curve is expected at some frequency higher than the dipole polarization peak.

These are generalizations that may not be the case with thermoplastic polymers with embedded fiber reinforcement as in the PEEK/AS4 carbon-fiber system being investigated. The peaks for dipole and ionic polarization may fall within the same elevated frequency range and be difficult to distinguish. Therefore, dielectric properties may be very different for pure polymers and for polymers with fibrous reinforcement. Processing reagents and surface treatment sizings

used on fibers to enhance fiber-polymer bonding can contribute ionic functional groups that would serve to increase the ionic polarizability of the polymeric regions.

Takahagi and Ishitani [9] attempted to characterize carbon-fiber surface chemistry. Their research showed a predominance of -OH and C=O functional groups, even without surface treatment. If the loss tangents of any associated molecules caused peaks at or near the same frequency as the dipole polarization inherent in the PEEK polymer, an effective increase in the loss tangent at that frequency would be seen.

Measuring this expected difference in dielectric properties between the neat PEEK polymer and the PEEK polymer with fiber reinforcement is not possible without first removing the fibers from the polymer due to the extreme difference in electrical properties between the polymer and the conductive carbon fibers. Our study of induction heating includes only the dielectric data obtained for virgin PEEK.

Both the dielectric constant and the dissipation factor are related to the presence of moisture. Moisture, absorbed in a filler at interfaces between different components or in laminations, can often be detected by changes in dissipation factor and capacitance [6]. Water molecules are highly polar and highly mobile. At frequencies of 10^5-10^6 Hz, the losses in water are ionic and increase with increases in dissolved ionizable salts. Hartshorn et al. [10] provide loss tangent vs. frequency data for several polymers in wet and dry states. The wet specimens, as a rule, gave higher loss tangent data at all frequencies. Potthoff [11] showed that increasing the temperature also increased the total effect of the presence of moisture in the system.

Moisture content studies on 150G PEEK [3] and PEEK and polyetherimide (PEI)-based composites [4, 5] indicate relatively low moisture contents in these advanced thermoplastic systems; saturated moisture absorption in PEEK-based APC-2 averaged about 0.2 weight-percent at room temperature and 100% relative humidity. Comparatively, the amorphous PES-based RADEL-C/T300 absorbed about 2.5% at equilibrium. The lower moisture absorption of PEEK is due to its semi-crystalline nature while the other resins (PEI and PES) are amorphous.

Buchman and Isayev [5] reported an equilibrium moisture content of about 0.6% for PEI-based CYPAC 7005/G30-500, while Demonet [4] reported 0.56%. The increase in moisture content from PEEK to PEI to PES may correspond to increases in loss tangent.

The next two sections describe the methods used to determine dielectric properties of PEEK over applicable ranges of frequency, temperature, and moisture pertinent to the present study and the results of those tests.

3. Characterization

The real and imaginary parts of the complex dielectric constant were measured at frequencies of 10³, 10⁴, 10⁵, and 10⁶ Hz using a cell built for similar characterization of filled polymers by Sturman [12, 13]. The physics of the cell and the specimen preparation is described by Sturman [12].

Specimens were cut from a semi-crystalline PEEK plate; the diameter of the machined disks was 4.01 cm and the thickness was 0.32 cm. A metal sleeve connected to a variac power supply surrounded the specimen. The power was incrementally increased and held for 15 min as current caused joule heating in the sleeve which conductively transmitted the energy to the polymer specimen. A thermocouple placed on the inside of the sleeve recorded the equilibrium temperature. Fifteen minutes between cycles allowed the aluminum sleeve, the cell electrodes, and the specimen to reach thermal equilibrium with the temperature increasing in increments of 25 °C from ambient to 250 °C.

The first set of measurements was performed at room temperature and at full moisture content (no drying). Results from three measurements were averaged, tabulated in Table 1 and shown in Figures 3 through 5 for the dielectric constant, loss tangent and resistivity versus frequency respectively. Figure 3 shows that the dielectric constant drops off rapidly between 100 kHz and 1 MHz. Figure 4 shows the expected increase in loss tangent with frequency in this

Table 1. Averaged Data From Three Electrical Property Measurements on PEEK at Room Temperature

Frequency (kHz)	Capacitance (pF)	Dielectric Constant	Loss Tangent	Conductance (1/ohm)	Resistance (ohm)	Resistivity (ohm-cm)
1	13.31	3.744	0.0014	1.00E - 10	1.00E + 10	4.02E + 11
10	13.282	3.736	0.0034	2.90E - 09	3.45E + 08	1.39E + 10
100	13.209	3.716	0.0160	1.33E - 07	7.52E + 06	3.02E + 08
1,000	12.392	3.486	0.0421	7.12E - 06	1.40E + 05	5.63E + 06

Note: The capacitance in vacuo is 3.555 pF for a surface area to diameter ratio of 0.402.

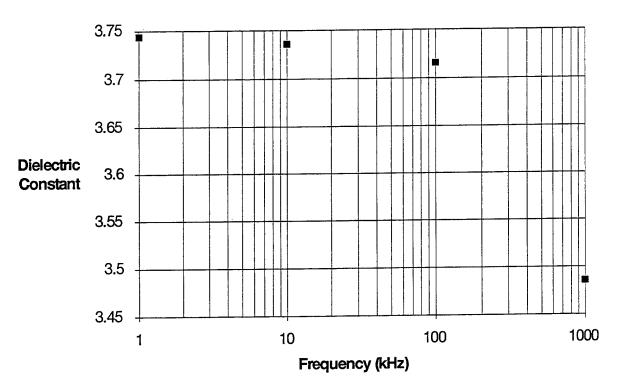


Figure 3. Room Temperature Dielectric Constant Results for PEEK.

range. The resistivity drops off linearly on the log-log plot of Figure 5 but is still of significant value at 1-MHz frequency.

These data are significantly higher in value in the higher frequency range than reported by PEEK's manufacturer, ICI [14]; ICI's data for 450G PEEK indicates loss tangents below 0.004

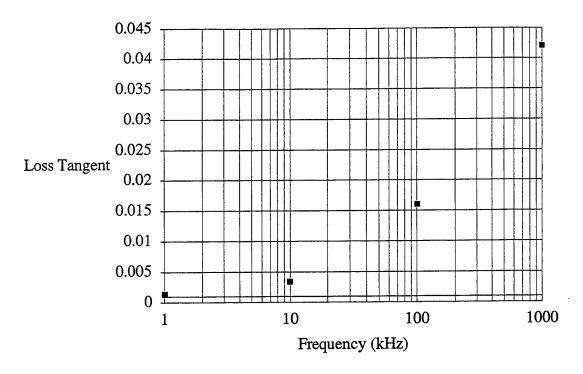


Figure 4. Room Temperature Dielectric Loss Tangent Results for PEEK.

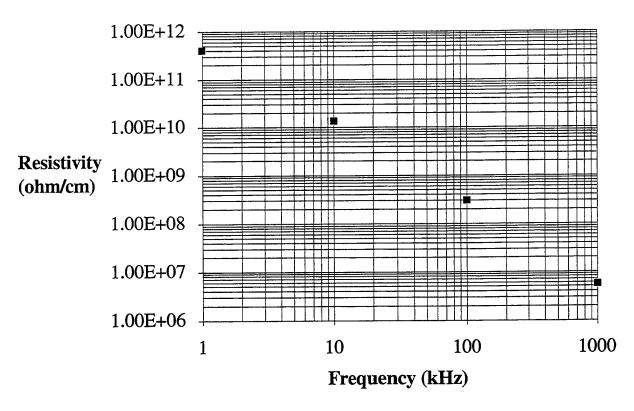


Figure 5. Room Temperature Results for Resistivity of PEEK.

at all frequencies between line frequency and 10^{10} Hz at room temperature. This discrepancy between our results and published data further precipitated an investigation of the effects of moisture content on dielectric loss tangent of PEEK.

Twelve specimens were cut from the original PEEK plate. Nine of the specimens were placed in an oven at 120 °C and 28-inHg vacuum in sets of three for periods of 24, 48, and 72 hr. The first three specimens were tested at temperatures ranging from 25 °C to 250 °C in 25 °C increments (nominally). They were assumed to have full equilibrium moisture content. The other specimens were tested as they were removed from the oven. Since each test took approximately 3 hr to run, some specimens were in the oven for 3–6 hr longer. Table 2 and Figure 6 show the dimensional and weight change data for all specimens. The solid squares represent the test data, while the line represents a spline fit to the average of the three data points at each nominal time plus 3 hr (which is the average time for each test set). From this plot, the maximum moisture content is taken to be 0.28 weight-percent; this value corresponds well to data reported in the literature.

Table 2. Weight Change Data for PEEK's Dielectric Loss Tangent vs. Moisture Content Testing

Time in Oven (hr)	Initial Weight (g)	Test Weight (g)	% Change	Average Change (%)
0	4.764	4.764	0	
0	4.812	4.812	0	0
0	4.693	4.693	0	
24	4.827	4.818	0.18	
27	4.71	4.701	0.19	0.1767
30	4.796	4.788	0.16	
48	4.774	4.762	0.25	
51	4.813	4.801	0.24	0.2533
54	4.682	4.669	0.27	
72	4.792	4.777	0.31	
75	4.845	4.832	0.27	0.28
78	4.613	4.601	0.26	

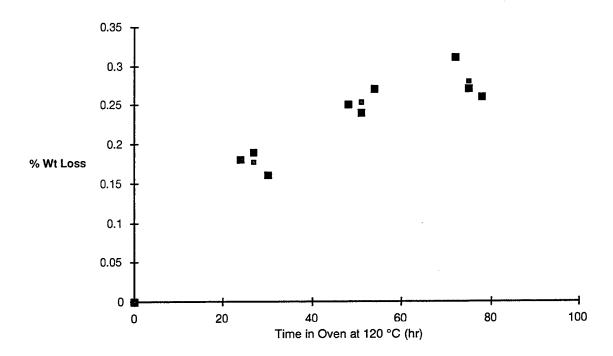


Figure 6. Percent Weight Loss of PEEK Specimens vs. Drying Time. Note That the Large Squares Represent the Three Specimens at Each Nominal Time. The Small Squares Represent the Average of the Three Data Points at Each Nominal Time.

Since maintaining the exact nominal temperatures was not possible, the results for the testing of each specimen have data points at different temperatures. However, the temperatures were generally maintained to within 5 °C of the nominal test temperature, so data has been averaged and plotted, assuming that all tests took place at the nominal test temperature. Figures 7 through 10 show the averaged loss tangent results for the 0-, 24-, 48-, and 72-hr specimens, respectively.

The influence of moisture content on the range of possible values of loss tangent at 100 kHz is shown in Figure 11. Here the 72-hr test is taken to be nearly zero moisture content and the 0-hr test is taken to be 0.28 weight-percent moisture. Also shown on Figure 11 is the value for loss tangent reported by ICI [14]. Although it was not reported as a function of moisture content, it might be concluded that their test specimens were indeed sufficiently dried prior to testing.

No attempt was made to quantify the percent moisture loss in fiber-reinforced PEEK. Furthermore, no attempt was made to increase the moisture content of the PEEK or APC-2

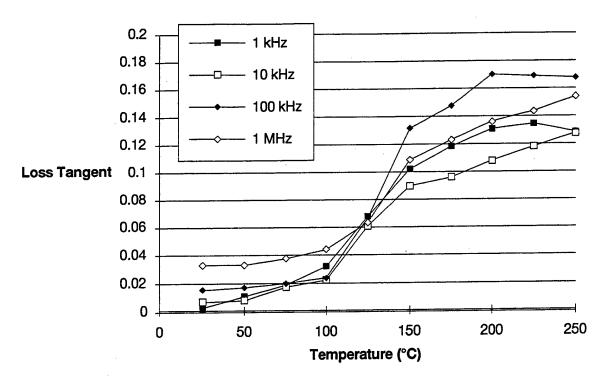


Figure 7. Average Experimental Loss Tangent Data for Three Undried PEEK Specimens.

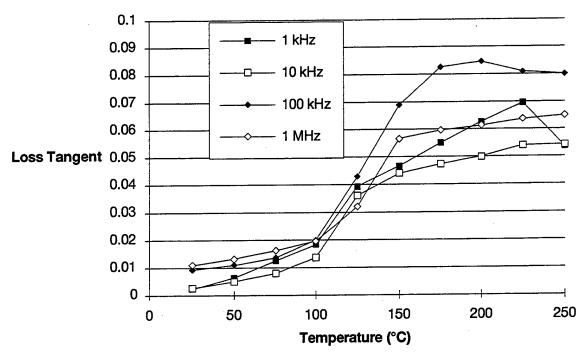


Figure 8. Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 24 hr.

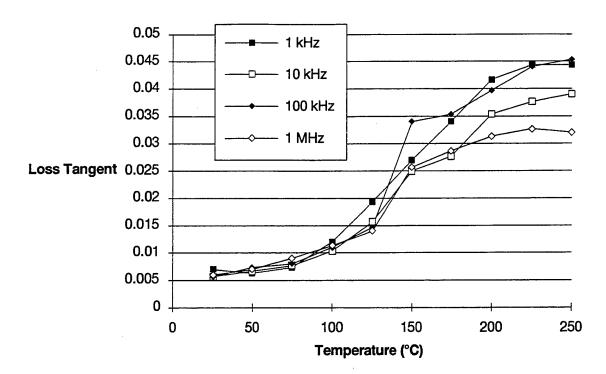


Figure 9. Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 48 hr.

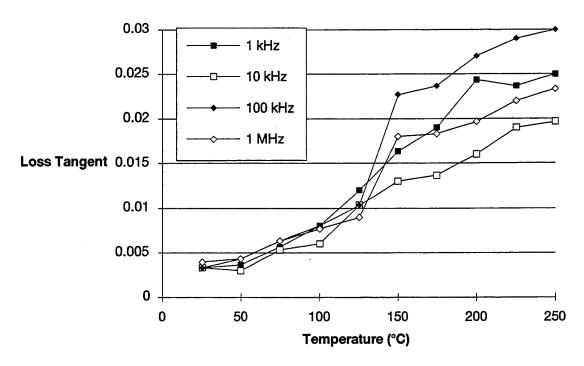


Figure 10. Average Experimental Loss Tangent Data for Three PEEK Specimens Dried at 120 °C for 72 hr.

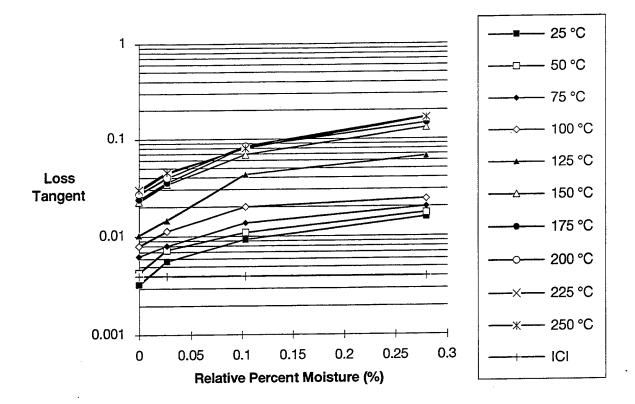


Figure 11. Experimental and ICI-Reported Loss Tangent Data for PEEK at 100 kHz. Note That the Experimental Data Approaches the ICI-Reported Data for Low Moisture Contents.

specimens, which, from the moisture dependence of loss tangent in Figure 10, might further increase heating. From this study, it appears possible to optimally localize heating at a bond interface in fusion bonding applications by increasing the moisture content of that local region; however, adding moisture in the presence of thermal gradients may nucleate undesireable voids in the composite [15].

4. Summary

Heating of continuous carbon-fiber-reinforced polymers (CFRP's) by the application of an alternating magnetic field has been shown to be due to dielectric losses in the polymer. A fundamental analysis indicated that heating in dielectric materials, such as PEEK, is linearly dependent upon the dielectric constant and the dissipation factor. Although not explicitly

characterized, induction heating in PEEK-based composites was shown to increase significantly with increases in moisture content and initial as-is loss tangent measurements showed relatively high values compared to those reported by the manufacturer. This precipitated a study of the effect of moisture content on dielectric properties. These properties were theoretically investigated as functions of temperature, frequency, ionic conductivity, and moisture content.

The imaginary part of the complex dielectric constant is an integral parameter in dielectric heating, and induction-heating thermal calculations are complicated by the loss tangent's dependence on frequency, temperature, moisture, and ionic conductivity. Fundamental tests on PEEK were performed to determine these relationships. A significant result is the extreme moisture dependence of the lossiness of PEEK. Under dry conditions, the loss data were near that reported by the manufacturer. Our immediate application of this study is to interpolate the expected loss tangent data for the frequency and temperature range expected in induction-heating experiments. This study has also established the importance of documenting the dependence of dielectric properties on frequency, temperature, and moisture.

From this study, it appears possible to optimally localize heating at a bond interface in fusion-bonding applications by increasing the moisture content of that local region; however, adding moisture in the presence of thermal gradients may nucleate undesireable voids in the composite [15].

INTENTIONALLY LEFT BLANK.

5. References

- 1. Fink, B. K., R. L. McCullough, and J. W. Gillespie Jr. "A Local Theory of Heating in Cross-Ply Carbon Fiber Thermoplastic Composites by Magnetic Induction." *Polymer Engineering and Science*, vol. 32, p. 357, 1992.
- 2. Fink, B. K. Dissertation. University of Delaware, 1991.
- 3. Pratte, J. F., W. H. Krueger, and I. Y. Chang, "High Performance Thermoplastic Composites with Poly(etheretherketone) Matrix." 34th International SAMPE Symposium Proceedings, 1989.
- 4. Demonet, C. M. "Interaction of Moisture With Resin Matrix Composites." 21st SAMPE Technical Conference Proceedings, 1989.
- 5. Buchman, A., and A. I. Isayev. "Water Absorption of Some Thermoplastic Composites." *SAMPE Journal*, vol. 27, 1991.
- 6. Mathes, K. Electrical and Electronic Properties of Polymers: A State-of-the-Art Compendum. J. I. Kroschwitz, (ed.), NY: John Wiley & Sons, Inc., 1988.
- 7. Von Hippel, A. Dielectric Materials and Applications. NY: The Technology Press of M.I.T. and John Wiley & Sons, Inc., 1966.
- 8. Ogorkiewicz, R. M. Engineering Properties of Thermoplastics. 1970.
- 9. Takahagi, T., and A. Ishitani. "XPS Studies by Use of the Digital Difference Spectrum Technique of Functional Groups on the Surface of CF." Carbon, vol. 22, pp. 41–46, 1984.
- 10. Hartshorn, L., J. Parry, and E. Rushton. Proc. Inst. Electr. Eng. Part 2. Vol. 100, p. 23, 1953.
- 11. Potthoff, K. Electrotech. Z. Vol. 80, p. 688, 1959.
- 12. Sturman, P. C., Jr. Doctoral dissertation. University of Delaware, 1990.
- 13. McCullough, R. L., and J. J. Kramer. "An Investigation of the Influence of Microstructure on the Electrical Properties of Composite Materials." Center for Composite Materials Report, University of Delaware, 1991.
- 14. ICI Advanced Materials. "Victrex PEEK 450G." ICI Advanced Materials, 1985.
- 15. Roychowdhury, S. Personal communication. 1991.

INTENTIONALLY LEFT BLANK.

NO. OF COPIES	ORGANIZATION	NO. OF <u>COPIES</u>	ORGANIZATION
2	DEFENSE TECHNICAL INFORMATION CENTER DTIC DDA 8725 JOHN J KINGMAN RD STE 0944 FT BELVOIR VA 22060-6218	1	DIRECTOR US ARMY RESEARCH LAB AMSRL D D R SMITH 2800 POWDER MILL RD ADELPHI MD 20783-1197
1	HQDA DAMO FDT 400 ARMY PENTAGON WASHINGTON DC 20310-0460	1	DIRECTOR US ARMY RESEARCH LAB AMSRL DD 2800 POWDER MILL RD ADELPHI MD 20783-1197
1	OSD OUSD(A&T)/ODDDR&E(R) R J TREW THE PENTAGON WASHINGTON DC 20301-7100	1	DIRECTOR US ARMY RESEARCH LAB AMSRL CS AS (RECORDS MGMT) 2800 POWDER MILL RD ADELPHI MD 20783-1145
1	DPTY CG FOR RDA US ARMY MATERIEL CMD AMCRDA 5001 EISENHOWER AVE ALEXANDRIA VA 22333-0001	3	DIRECTOR US ARMY RESEARCH LAB AMSRL CI LL 2800 POWDER MILL RD ADELPHI MD 20783-1145
	INST FOR ADVNCD TCHNLGY THE UNIV OF TEXAS AT AUSTIN PO BOX 202797 AUSTIN TX 78720-2797		ABERDEEN PROVING GROUND
_	DARPA B KASPAR 3701 N FAIRFAX DR ARLINGTON VA 22203-1714	. 4	DIR USARL AMSRL CI LP (BLDG 305)
	NAVAL SURFACE WARFARE CTR CODE B07 J PENNELLA 17320 DAHLGREN RD BLDG 1470 RM 1101 DAHLGREN VA 22448-5100		
	US MILITARY ACADEMY MATH SCI CTR OF EXCELLENCE DEPT OF MATHEMATICAL SCI MADN MATH THAYER HALL WEST POINT NY 10996-1786		

NO. OF COPIES	<u>ORGANIZATION</u>	NO. OF COPIES	ORGANIZATION
1	DIRECTOR US ARMY RESEARCH LAB AMSRL CP CA D SNIDER 2800 POWDER MILL RD ADELPHI MD 20783-1145	1	COMMANDER US ARMY MATERIEL CMD AMXMI INT 5001 EISENHOWER AVE ALEXANDRIA VA 22333-0001
1	DIRECTOR US ARMY RESEARCH LAB AMSRL OP SD TA 2800 POWDER MILL ROAD ADELPHI MD 20783-1145	2	COMMANDER US ARMY ARDEC AMSTA AR AE WW E BAKER J PEARSON PICATINNY ARSENAL NJ 07806-5000
3	DIRECTOR US ARMY RESEARCH LAB AMSRL OP SD TL 2800 POWDER MILL ROAD ADELPHI MD 20783-1145 DIRECTOR	1	COMMANDER US ARMY ARDEC AMSTA AR TD C SPINELLI PICATINNY ARSENAL NJ
1	US ARMY RESEARCH LAB AMSRL OP SD TP 2800 POWDER MILL ROAD ADELPHI MD 20783-1145	1	07806-5000 COMMANDER US ARMY ARDEC AMSTA AR FSE T GORA
2	DIRECTOR US ARMY RESEARCH LAB AMSRL OP CI AD TECH PUB BR RECORDS MGMT ADMIN 2800 POWDER MILL ROAD ADELPHI MD 20783-1197	6	PICATINNY ARSENAL NJ COMMANDER US ARMY ARDEC AMSTA AR CCH A W ANDREWS S MUSALLI
1	HQDA DAMI FIT NOLAN BLDG WASHINGTON DC 20310-1025 DIRECTOR		R CARR M LUCIANO E LOGSDEN T LOUZEIRO PICATINNY ARSENAL NJ 07806-5000
	DA OASARDA SARD SO 103 ARMY PENTAGON WASHINGTON DC 20310-0103	4	COMMANDER US ARMY ARDEC AMSTA AR CC G PAYNE J GEHBAUER
1	DEPUTY ASST SCY FOR R&T SARD TT RM 3EA79 THE PENTAGON WASHINGTON DC 20301-7100		C BAULIEU H OPAT PICATINNY ARSENAL NJ 07806-5000

NO. OF COPIES	<u>ORGANIZATION</u>	NO. OF <u>COPIES</u>	<u>ORGANIZATION</u>
1	COMMANDER US ARMY ARDEC AMSTA AR CCH P J LUTZ PICATINNY ARSENAL NJ 07806-5000		COMMANDER US ARMY ARDEC AMSTA AR FSP A P KISATSKY PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR FSF T C LIVECCHIA PICATINNY ARSENAL NJ 07806-5000	2	COMMANDER US ARMY ARDEC AMSTA AR CCH C H CHANIN S CHICO PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR QAC T C C PATEL PICATINNY ARSENAL NJ 07806-5000	9	COMMANDER US ARMY ARDEC AMSTA AR CCH B P DONADIA F DONLON P VALENTI
2	COMMANDER US ARMY ARDEC AMSTA AR M D DEMELLA F DIORIO PICATINNY ARSENAL NJ 07806-5000 COMMANDER		C KNUTSON G EUSTICE S PATEL G WAGNECZ R SAYER F CHANG PICATINNY ARSENAL NJ 07806-5000
J	US ARMY ARDEC AMSTA AR FSA A WARNASH B MACHAK M CHIEFA PICATINNY ARSENAL NJ 07806-5000	6	COMMANDER US ARMY ARDEC AMSTA AR CCL F PUZYCKI R MCHUGH D CONWAY E JAROSZEWSKI R SCHLENNER
2	COMMANDER US ARMY ARDEC AMSTA AR FSP G M SCHIKSNIS D CARLUCCI PICATINNY ARSENAL NJ 07806-5000	1	M CLUNE PICATINNY ARSENAL NJ 07806-5000 COMMANDER US ARMY ARDEC AMSTA AR QAC T D RIGOGLIOSO PICATINNY ARSENAL NJ 07806-5000

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
1	COMMANDER US ARMY ARDEC AMSTA AR SRE D YEE PICATINNY ARSENAL NJ 07806-5000	2	PEO FIELD ARTILLERY SYSTEMS SFAE FAS PM H GOLDMAN T MCWILLIAMS PICATINNY ARSENAL NJ 07806-5000
1	COMMANDER US ARMY ARDEC AMSTA AR WET T SACHAR BLDG 172 PICATINNY ARSENAL NJ 07806-5000 COMMANDER US ARMY ARDEC	6	PM SADARM SFAE GCSS SD COL B ELLIS M DEVINE R KOWALSKI W DEMASSI J PRITCHARD S HROWNAK PICATINNY ARSENAL NJ 07806-5000
	SMCAR ASF PICATINNY ARSENAL NJ 07806-5000	1	COMMANDER US ARMY ARDEC PRODUCTION BASE
1	COMMANDER US ARMY ARDEC AMSTA AR WEL F INTELLIGENCE SPECIALIST M GUERRIERE		MODERN ACTY AMSMC PBM K PICATINNY ARSENAL NJ 07806-5000
	PICATINNY ARSENAL NJ 07806-5000	3	COMMANDER U S ARMY TACOM PM TACTICAL VEHICLES
11	PROJECT MANAGER TANK MAIN ARMAMENT SYSTEMS SFAE GSSC TMA R MORRIS C KIMKER		SFAE TVL SFAE TVM SFAE TVH 6501 ELEVEN MILE RD WARREN MI 48397-5000
	D GUZOWICZ E KOPACZ R ROESER R DARCY R MCDANOLDS L D ULISSE	1	COMMANDER U S ARMY TACOM PM ABRAMS SFAE ASM AB 6501 ELEVEN MILE RD WARREN MI 48397-5000
	C ROLLER J MCGREEN B PATTER PICATINNY ARSENAL NJ 07806-5000	1	COMMANDER U S ARMY TACOM PM BFVS SFAE ASM BV 6501 ELEVEN MILE RD WARREN MI 48397-5000

NO. OF COPIES	<u>ORGANIZATION</u>	NO. OF COPIES	ORGANIZATION
1	COMMANDER U S ARMY TACOM PM AFAS SFAE ASM AF 6501 ELEVEN MILE RD WARREN MI 48397-5000	1	COMMANDER U S ARMY TACOM PM GROUND SYSTEMS INTEGRATION SFAE GCSS W GSI R LABATILLE 6501 ELEVEN MILE RD
2	COMMANDER U S ARMY TACOM PM SURV SYS SFAE ASM SS T DEAN SFAE GCSS W GSI M D COCHRAN 6501 ELEVEN MILE RD WARREN MI 48397-5000	1	WARREN MI 48397-5000 COMMANDER U S ARMY TACOM CHIEF ABRAMS TESTING SFAE GCSS W AB QT T KRASKIEWICZ 6501 ELEVEN MILE RD WARREN MI 48397-5000
	COMMANDER U S ARMY TACOM PM RDT&E SFAE GCSS W AB J GODELL 6501 ELEVEN MILE RD	1	COMMANDER US ARMY TACOM AMSTA SF WARREN MI 48397-5000 COMMANDER
1	WARREN MI 48397-5000 COMMANDER U S ARMY TACOM PM SURVIVABLE SYSTEMS SFAE GCSS W GSI H	·	SMCWV QAE Q B VANINA BLDG 44 WATERVLIET ARSENAL WATERVLIET NY 12189-4050
	M RYZYI 6501 ELEVEN MILE RD WARREN MI 48397-5000	14	COMMANDER US ARMY TACOM ASMTA TR R J CHAPIN
1	COMMANDER U S ARMY TACOM PM BFV SFAE GCSS W BV S DAVIS 6501 ELEVEN MILE RD WARREN MI 48397-5000		R MCCLELLAND D THOMAS J BENNETT D HANSEN AMSTA JSK S GOODMAN J FLORENCE K IYER
1	COMMANDER U S ARMY TACOM PM LIGHT TACTICAL VEHICLES AMSTA TR S AJ J MILLS MS 209 6501 ELEVEN MILE RD WARREN MI 48397-5000		J THOMSON AMSTA TR D D OSTBERG L HINOJOSA B RAJU AMSTA CS SF H HUTCHINSON F SCHWARZ WARREN MI 48397-5000

NO. OF COPIES	<u>ORGANIZATION</u>	NO. OF COPIES	ORGANIZATION
1	COMMANDER SMCWV SPM T MCCLOSKEY BLDG 253 WATERVLIET ARSENAL WATERVLIET NY 12189-4050	4	DIRECTOR US ARMY CECOM NIGHT VISION AND ELECTRONIC SENSORS DIRECTORATE AMSEL RD NV CM CCD R ADAMS R MCLEAN
10	BENET LABORATORIES AMSTA AR CCB R FISCELLA G D ANDREA M SCAVULO G SPENCER		A YINGST AMSEL RD NV VISP E JACOBS 10221 BURBECK RD FT BELVOIR VA 22060-5806
	P WHEELER K MINER J VASILAKIS G FRIAR R HASENBEIN	2	CDR USA AMCOM AVIATION APPLIED TECH DIR J SCHUCK FT EUSTIS VA 23604-5577
	SMCAR CCB R S SOPOK WATER VLIET NY 12189	1	U S ARMY CRREL P DUTTA 72 LYME RD HANOVER NH 03755
2	TSM ABRAMS ATZK TS S JABURG W MEINSHAUSEN FT KNOX KY 40121	1	US ARMY CERL R LAMPO 2902 NEWMARK DR CHAMPAIGN IL 61822
3	ARMOR SCHOOL ATTN ATZK TD R BAUEN J BERG A POMEY FT KNOX KY 40121	2	U S ARMY CORP OF ENGINEERS CERD C T LIU CEW ET T TAN 20 MASS AVE NW WASHINGTON DC 20314
2	HQ IOC TANK AMMO TEAM AMSIO SMT R CRAWFORD W HARRIS ROCK ISLAND IL 61299-6000	10	DIRECTOR US ARMY NATL GRND INTEL CTR D LEITER S EITELMAN M HOLTUS M WOLFE S MINGLEDORF
1	DIRECTOR U S ARMY AMCOM SFAE AV RAM TV D CALDWELL BUILDING 5300 REDSTONE ARSENAL AL 35898		H C ARDLEIGH J GASTON W GSTATTENBAUER R WARNER J CRIDER 220 SEVENTH STREET NE CHARLOTTESVILLE VA 22091

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
6	US ARMY SBCCOM SOLDIER SYSTEMS CENTER BALLISTICS TEAM J WARD MARINE CORPS TEAM J MACKIEWICZ	1	COMMANDANT U S ARMY FIELD ARTILLERY CTR AT FT SILL ATFS CD LTC BUMGARNER FT SILL OK 73503 5600
	BUS AREA ADVOCACY TEAM W HASKELL SSCNC WST W NYKVIST T MERRILL	1	CHIEF USAIC LTC T J CUMMINGS ATZB COM FT BENNING GA 31905-5800
	S BEAUDOIN KANSAS ST NATICK MA 01760-5019	1	NAVAL AIR SYSTEMS CMD J THOMPSON 48142 SHAW RD UNIT 5 PATUXENT RIVER MD 20670
1	US ARMY COLD REGIONS RSCH & ENGRNG LAB P DUTTA 72 LYME RD	1	NAVAL SURFACE WARFARE CTR DAHLGREN DIV CODE G06 DAHLGREN VA 22448
	HANOVER NH 03755	1	NAVAL SURFACE WARFARE CTR TECH LIBRARY CODE 323
1	SYSTEM MANAGER ABRAMS BLDG 1002 RM 110 ATZK TS LTC J H NUNN		17320 DAHLGREN RD DAHLGREN VA 22448
9	FT KNOX KY 40121 US ARMY RESEARCH OFFICE A CROWSON J CHANDRA	3	NAVAL RESEARCH LAB I WOLOCK CODE 6383 R BADALIANCE CODE 6304 L GAUSE WASHINGTON DC 20375
	H EVERETT J PRATER R SINGLETON G ANDERSON D STEPP D KISEROW	1	NAVAL SURFACE WARFARE CTR CRANE DIVISION M JOHNSON CODE 20H4 LOUISVILLE KY 40214-5245
	J CHANG PO BOX 12211 RESEARCH TRIANGLE PARK NC 27709-2211	2	COMMANDER NAVAL SURFACE WARFARE CTR CADEROCK DIVISION R PETERSON CODE 2020 M CRITCHFIELD CODE 1730
1	DIRECTORATE OF CMBT DEVELOPMENT C KJORO 320 ENGINEER LOOP STE 141 FT LEONARD WOOD MO 65473-8929	2	NAVAL SURFACE WARFARE CTR U SORATHIA C WILLIAMS CD 6551 9500 MACARTHUR BLVD WEST BETHESDA MD 20817

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
1 .	DAVID TAYLOR RESEARCH CTR SHIP STRUCTURES & PROTECTION DEPARTMENT CODE 1702 J CORRADO BETHESDA MD 20084	1	NAVSEA OJRI PEO DD21 PMS500 G CAMPONESCHI 2351 JEFFERSON DAVIS HWY ARLINGTON VA 22242-5165
2	DAVID TAYLOR RESEARCH CTR R ROCKWELL W PHYILLAIER BETHESDA MD 20054-5000	1	EXPEDITIONARY WARFARE DIV N85 F SHOUP 2000 NAVY PENTAGON WASHINGTON DC 20350-2000
1	OFFICE OF NAVAL RESEARCH D SIEGEL CODE 351 800 N QUINCY ST ARLINGTON VA 22217-5660	1	AFRL MLBC 2941 P STREET RM 136 WRIGHT PATTERSON AFB OH 45433-7750
8	NAVAL SURFACE WARFARE CTR J FRANCIS CODE G30 D WILSON CODE G32 R D COOPER CODE G32 J FRAYSSE CODE G33 E ROWE CODE G33	1	AFRL MLSS R THOMSON 2179 12TH STREET RM 122 WRIGHT PATTERSON AFB OH 45433-7718
	T DURAN CODE G33 L DE SIMONE CODE G33 R HUBBARD CODE G33 DAHLGREN VA 22448	2	AFRL F ABRAMS J BROWN BLDG 653
1	NAVAL SEA SYSTEMS CMD D LIESE 2531 JEFFERSON DAVIS HIGHWAY ARLINGTON VA 22242-5160		2977 P STREET STE 6 WRIGHT PATTERSON AFB OH 45433-7739
1	NAVAL SURFACE WARFARE M LACY CODE B02 17320 DAHLGREN RD DAHLGREN VA 22448	1	AFRL MLS OL 7278 4TH STREET BLDG 100 BAY D L COULTER HILL AFB UT 84056-5205
1	OFFICE OF NAVAL RES J KELLY 800 NORTH QUINCEY ST ARLINGTON VA 22217-5000	1	OSD JOINT CCD TEST FORCE OSD JCCD R WILLIAMS 3909 HALLS FERRY RD VICKSBURG MS 29180-6199
2	NAVAL SURFACE WARFARE CTR CARDEROCK DIVISION R CRANE CODE 2802 C WILLIAMS CODE 6553 3A LEGGETT CIR BETHESDA MD 20054-5000	1	DEFENSE NUCLEAR AGENCY INNOVATIVE CONCEPTS DIV R ROHR 6801 TELEGRAPH RD ALEXANDRIA VA 22310-3398

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	<u>ORGANIZATION</u>
1	WATERWAYS EXPERIMENT D SCOTT 3909 HALLS FERRY RD SC C VICKSBURG MS 39180	5	DIRECTOR LAWRENCE LIVERMORE NATL LAB R CHRISTENSEN S DETERESA
3	DARPA M VANFOSSEN S WAX L CHRISTODOULOU 3701 N FAIRFAX DR ARLINGTON VA 22203-1714	_	F MAGNESS M FINGER MS 313 M MURPHY L 282 PO BOX 808 LIVERMORE CA 94550
2	SERDP PROGRAM OFC PM P2 C PELLERIN B SMITH 901 N STUART ST SUITE 303 ARLINGTON VA 22203	7	NIST R PARNAS J DUNKERS M VANLANDINGHAM MS 8621 J CHIN MS 8621 D HUNSTON MS 8543 J MARTIN MS 8621
1	FAA MIL HDBK 17 CHAIR L ILCEWICZ 1601 LIND AVE SW ANM 115N RENTON VA 98055	1	D DUTHINH MS 8611 100 BUREAU DR GAITHERSBURG MD 20899 OAK RIDGE NATL LAB C EBERLE MS 8048 PO BOX 2009
2	FAA TECH CENTER D OPLINGER AAR 431 P SHYPRYKEVICH AAR 431 ATLANTIC CITY NJ 08405	1	OAK RIDGE TN 37831 OAK RIDGE NATL LAB C D WARREN MS 8039
1	OFC OF ENVIRONMENTAL MGMT U S DEPT OF ENERGY		PO BOX 2009 OAK RIDGE TN 37922
	P RITZCOVAN 19901 GERMANTOWN RD GERMANTOWN MD 20874-1928	4	DIRECTOR SANDIA NATL LABS APPLIED MECHANICS DEPT DIVISION 8241
1	LOS ALAMOS NATL LAB F ADDESSIO		W KAWAHARA K PERANO
	MS B216 PO BOX 1633 LOS ALAMOS NM 87545		D DAWSON P NIELAN PO BOX 969 LIVERMORE CA 94550-0096
1	OAK RIDGE NATL LAB R M DAVIS PO BOX 2008 OAK RIDGE TN 37831-6195	1	LAWRENCE LIVERMORE NATIONAL LAB M MURPHY PO BOX 808 L 282 LIVERMORE CA 94550

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
3	NASA LANGLEY RESEARCH CTR MS 266 AMSRL VS	. 1	GRAPHITE MASTERS INC J WILLIS 3815 MEDFORD ST
	W ELBER F BARTLETT JR G FARLEY	1	LOS ANGELES CA 90063-1900 ADVANCED GLASS FIBER YARNS
1	HAMPTON VA 23681-0001 NASA LANGLEY RESEARCH CTR		T COLLINS 281 SPRING RUN LN STE A DOWNINGTON PA 19335
1	T GATES MS 188E HAMPTON VA 23661-3400	1	COMPOSITE MATERIALS INC
1	USDOT FEDERAL RAILROAD		D SHORTT 19105 63 AVE NE
	RDV 31 M FATEH WASHINGTON DC 20590		PO BOX 25 ARLINGTON WA 98223
1	DOT FHWA J SCALZI	1	COMPOSITE MATERIALS INC R HOLLAND
	400 SEVENTH ST SW 3203 HNG 32 WASHINGTON DC 20590		11 JEWEL COURT ORINDA CA 94563
1	FHWA	1	COMPOSITE MATERIALS INC
	E MUNLEY 6300 GEORGETOWN PIKE MCLEAN VA 22101		14530 S ANSON AVE SANTA FE SPRINGS CA 90670
1	CENTRAL INTELLIGENCE AGENCY	2	COMPOSIX D BLAKE L DIXON
	OTI WDAG GT W L WALTMAN PO BOX 1925		120 O NEILL DR HEBRUN OHIO 43025
	WASHINGTON DC 20505 MARINE CORPS INTEL ACTY	4	CYTEC FIBERITE R DUNNE
1	D KOSITZKE 3300 RUSSELL RD SUITE 250 QUANTICO VA 22134-5011		D KOHLI M GILLIO R MAYHEW 1300 REVOLUTION ST
1	NATL GRND INTELLIGENCE CTR DIRECTOR		HAVRE DE GRACE MD 21078
	IANG TMT 220 SEVENTH ST NE CHARLOTTESVILLE VA 22902-5396	2	SIMULA J COLTMAN R HUYETT 10016 S 51ST ST
1	DIRECTOR DEFENSE INTELLIGENCE AGENCY TA 5 K CRELLING	1	PHOENIX AZ 85044 SIOUX MFG
	WASHINGTON DC 20310	*	B KRIEL PO BOX 400 FT TOTTEN ND 58335

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
2	PROTECTION MATERIALS INC M MILLER F CRILLEY 14000 NW 58 CT MIAMI LAKES FL 33014	1	BATTELLE C R HARGREAVES 505 KING AVE COLUMBUS OH 43201-2681
	MAMI LAKES I'E 35014	2	BATTELLE NATICK OPERATIONS
3	FOSTER MILLER		J CONNORS
	J J GASSNER		B HALPIN
	M ROYLANCE		209 W CENTRAL ST
	W ZUKAS		STE 302 NATICK MA 01760
	195 BEAR HILL RD WALTHAM MA 02354-1196		NATICA MA 01/00
	WALIHAM MA 02354-1190	1	BATTELLE NW DOE PNNL
1	ROM DEVELOPMENT CORP	1	T HALL MS K231
1	R O MEARA		BATTELLE BLVD
	136 SWINEBURNE ROW		RICHLAND WA 99352
	BRICK MARKET PLACE		
	NEWPORT RI 02840	3	PACIFIC NORTHWEST LAB
			M SMITH
2	TEXTRON SYSTEMS		G VAN ARSDALE
	T FOLTZ		R SHIPPELL
	M TREASURE		PO BOX 999
	201 LOWELL ST		RICHLAND WA 99352
	WILMINGTON MA 08870-2941	1	ARMTEC DEFENSE PRODUCTS
1	JPS GLASS	1	S DYER
1	L CARTER		85 901 AVE 53
	PO BOX 260		PO BOX 848
	SLATER RD		COACHELLA CA 92236
	SLATER SC 29683		
		2	ADVANCED COMPOSITE
1	O GARA HESS & EISENHARDT		MATLS CORP
	M GILLESPIE		P HOOD
	9113 LESAINT DR FAIRFIELD OH 45014		J RHODES 1525 S BUNCOMBE RD
	PAIRFIELD ON 45014		GREER SC 29651-9208
2	MILLIKEN RESEARCH CORP		QIADAK OC 27001 7200
-	H KUHN	2	GLCC INC
	M MACLEOD		JRAY
	PO BOX 1926		M BRADLEY
	SPARTANBURG SC 29303		103 TRADE ZONE DR
			STE 26C
1	CONNEAUGHT INDUSTRIES INC J SANTOS		WEST COLUMBIA SC 29170
	PO BOX 1425	2	AMOCO PERFORMANCE
	COVENTRY RI 02816		PRODUCTS
			M MICHNO JR
			J BANISAUKAS 4500 MCGINNIS FERRY RD
			ALPHARETTA GA 30202-3944
			ALI HAICH IA OA 30202-3744

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
1	SAIC M PALMER 2109 AIR PARK RD S E	1	PROJECTILE TECHNOLOGY INC 515 GILES ST HAVRE DE GRACE MD 21078
1	ALBUQUERQUE NM 87106 SAIC	1	CUSTOM ANALYTICAL ENG SYS INC
-	ATTN G CHRYSSOMALLIS 3800 W 80TH ST STE 1090 BLOOMINGTON MN 55431		A ALEXANDER 13000 TENSOR LN NE FLINTSTONE MD 21530
1	AAI CORPORATION DR T G STASTNY PO BOX 126	2	LORAL VOUGHT SYSTEMS G JACKSON K COOK
1	HUNT VALLEY MD 21030-0126 JOHN HEBERT		1701 W MARSHALL DR GRAND PRAIRIE TX 75051
1	PO BOX 1072 HUNT VALLEY MD 21030-0126	5	AEROJET GEN CORP D PILLASCH T COULTER
12	ALLIANT TECHSYSTEMS INC C CANDLAND C AAKHUS R BECKER		C FLYNN D RUBAREZUL M GREINER 1100 WEST HOLLYVALE ST AZUSA CA 91702-0296
	B SEE N VLAHAKUS R DOHRN	3	HEXCEL INC
	S HAGLUND D FISHER W WORRELL R COPENHAFER M HISSONG		R BOE F POLICELLI J POESCH PO BOX 98 MAGNA UT 84044
	D KAMDAR 600 2ND ST NE HOPKINS MN 55343-8367	3	HERCULES INC G KUEBELER J VERMEYCHUK
3	ALLIANT TECHSYSTEMS INC J CONDON E LYNAM		B MANDERVILLE JR HERCULES PLAZA WILMINGTON DE 19894
	J GERHARD WV01 16 STATE RT 956 PO BOX 210 ROCKET CENTER WV 26726-0210	1	BRIGS COMPANY J BACKOFEN 2668 PETERBOROUGH ST HERDON VA 22071-2443
1	APPLIED COMPOSITES W GRISCH 333 NORTH SIXTH ST ST CHARLES IL 60174	1	ZERNOW TECHNICAL SERVICES L ZERNOW 425 W BONITA AVE STE 208 SAN DIMAS CA 91773

NO. OF COPIES	<u>ORGANIZATION</u>	NO. OF COPIES	<u>ORGANIZATION</u>
2	OLIN CORPORATION	1	BOEING
_	FLINCHBAUGH DIV	_	R BOHLMANN
,	E STEINER		PO BOX 516 MC 5021322
	B STEWART		ST LOUIS MO 63166-0516
	PO BOX 127		
	RED LION PA 17356	2	BOEING DEFENSE
			AND SPACE GRP
1	OLIN CORPORATION		W HAMMOND
	L WHITMORE		J RUSSELL
	10101 9TH ST NORTH		S 4X55
•	ST PETERSBURG FL 33702		PO BOX 3707
_			SEATTLE WA 98124-2207
1	DOWUT	•	DOEDIG DOMODOD A FEE
	S TIDRICK	2	BOEING ROTORCRAFT
	15 STERLING DR		P MINGURT P HANDEL
•	WALLINGFORD CT 06492		800 B PUTNAM BLVD
5	SIKORSKY AIRCRAFT		WALLINGFORD PA 19086
3	G JACARUSO		WALLINGFORD PA 19080
	T CARSTENSAN	1	BOEING
	B KAY	1	DOUGLAS PRODUCTS DIV
	S GARBO M S S330A		L J HART SMITH
	J ADELMANN		3855 LAKEWOOD BLVD
	6900 MAIN ST		D800 0019
	PO BOX 9729		LONG BEACH CA 90846-0001
	STRATFORD CT 06497-9729		
		1	LOCKHEED MARTIN
1	PRATT & WHITNEY		S REEVE
	D HAMBRICK		8650 COBB DR
	400 MAIN ST MS 114 37		D 73 62 MZ 0648
	EAST HARTFORD CT 06108		MARIETTA GA 30063-0648
1	AEROSPACE CORP	1	LOCKHEED MARTIN
	G HAWKINS M4 945		SKUNK WORKS
	2350 E EL SEGUNDO BLVD		D FORTNEY
	EL SEGUNDO CA 90245		1011 LOCKHEED WAY
			PALMDALE CA 93599-2502
2	CYTEC FIBERITE		
	M LIN	1	LOCKHEED MARTIN
	W WEB		R FIELDS
	1440 N KRAEMER BLVD		1195 IRWIN CT
	ANAHEIM CA 92806		WINTER SPRINGS FL 32708
1	HEXCEL	1	MATERIALS SCIENCES CORP
	T BITZER		B W ROSEN
	11711 DUBLIN BLVD		500 OFFICE CENTER DR STE 250
	DUBLIN CA 94568		FORT WASHINGTON PA 19034

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
1	NORTHRUP GRUMMAN CORP ELECTRONIC SENSORS & SYSTEMS DIV E SCHOCH 1745A WEST NURSERY RD MAILSTOP V 16 LINTHICUM MD 21090	2	GENERAL DYNAMICS LAND SYSTEMS D REES M PASIK PO BOX 2074 WARREN MI 48090-2074
2	NORTHROP GRUMMAN ENVIRONMENTAL PROGRAMS R OSTERMAN A YEN 8900 E WASHINGTON BLVD PICO RIVERA CA 90660	1	GENERAL DYNAMICS LAND SYSTEMS DIVISION D BARTLE PO BOX 1901 WARREN MI 48090 GENERAL DYNAMICS
1	UNITED DEFENSE LP PO BOX 359 D MARTIN SANTA CLARA CA 95052		LAND SYSTEMS MUSKEGON OPERATIONS W SOMMERS JR 76 GETTY ST MUSKEGON MI 49442
1	UNITED DEFENSE LP PO BOX 58123 G THOMAS SANTA CLARA CA 95052	1	GENERAL DYNAMICS AMPHIBIOUS SYS SURVIVABILITY LEAD G WALKER 991 ANNAPOLIS WAY
2	UNITED DEFENSE LP MAIL DROP M53 R BARRETT V HORVATICH 328 W BROKAW RD SANTA CLARA CA 95052-0359	5	WOODBRIDGE VA 22191 INSTITUTE FOR ADVANCED TECH T KIEHNE H FAIR P SULLIVAN
3	UNITED DEFENSE LP GROUND SYSTEMS DIVISION M PEDRAZZI MAIL DROP N09 A LEE MAIL DROP N11 M MACLEAN MAIL DROP N06 1205 COLEMAN AVE	2	W REINECKE I MCNAB 4030 2 W BRAKER LN AUSTIN TX 78759 CIVIL ENGR RSCH FOUNDATION H DEPOSITED PRESIDENT
4	SANTA CLARA CA 95052 UNITED DEFENSE LP 4800 EAST RIVER RD R BRYNSVOLD		H BERNSTEIN PRESIDENT R BELLE 1015 15TH ST NW STE 600 WASHINGTON DC 20005 ARROW TECH ASSO
	P JANKE MS170 T GIOVANETTI MS236 B VAN WYK MS389 MINNEAPOLIS MN 55421-1498	1	1233 SHELBURNE RD STE D 8 SOUTH BURLINGTON VT 05403-7700

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
1	CONSULTANT R EICHELBERGER 409 W CATHERINE ST BEL AIR MD 21014-3613	1	UNIVERSITY OF UTAH DEPT OF MECH & INDUSTRIAL ENGR S SWANSON SALT LAKE CITY UT 84112
1	UCLA MANE DEPT ENGR IV H THOMAS HAHN LOS ANGELES CA 90024-1597	2	PENNSYLVANIA STATE UNIV R MCNITT C BAKIS
2	U OF DAYTON RESEARCH INSTUTE RAN Y KIM AJIT K ROY		227 HAMMOND BLDG UNIVERSITY PARK PA 16802
	300 COLLEGE PARK AVE DAYTON OH 45469-0168	1	PENNSYLVANIA STATE UNIV RENATA S ENGEL 245 HAMMOND BLDG
1	MIT PLAGACE	1	UNIVERSITY PARK PA 16801 PURDUE UNIVERSITY
	77 MASS AVE CAMBRIDGE MA 01887	1	SCHOOL OF AERO & ASTRO C T SUN
1	IIT RESEARCH CENTER D ROSE	_	W LAFAYETTE IN 47907-1282
	201 MILL ST ROME NY 13440-6916	1	STANFORD UNIVERSITY DEPARTMENT OF AERONAUTICS AND AEROBALLISTICS
1	GEORGIA TECH RESEARCH INSTITUTE GEORGIA INSTITUTE		DURANT BUILDING S TSAI STANFORD CA 94305
	OF TECHNOLOGY P FRIEDERICH ATLANTA GA 30392	1	UNIVERSITY OF DAYTON J M WHITNEY
1	MICHIGAN ST UNIVERSITY R AVERILL		COLLEGE PARK AVE DAYTON OH 45469-0240
	3515 EB MSM DEPT EAST LANSING MI 48824-1226	7	UNIVERSITY OF DELAWARE CTR FOR COMPOSITE MATERIALS
1	UNIVERSITY OF KENTUCKY LYNN PENN 763 ANDERSON HALL LEXINGTON KY 40506-0046		J GILLESPIE M SANTARE G PALMESE S YARLAGADDA
1	UNIVERSITY OF WYOMING D ADAMS PO BOX 3295 LARAMIE WY 82071		S ADVANI D HEIDER D KUKICH 201 SPENCER LABORATORY NEWARK DE 19716
			-

NO. OF COPIES	ORGANIZATION	NO. OF COPIES	ORGANIZATION
1	UNIVERSITY OF ILLINOIS AT URBANA CHAMPAIGN NATL CENTER FOR COMPOSITE MATERIALS RESEARCH 216 TALBOT LABORATORY J ECONOMY 104 S WRIGHT ST URBANA IL 61801	. 1	ABERDEEN PROVING GROUND COMMANDER US ARMY MATERIEL SYS ANALYSIS P DIETZ 392 HOPKINS RD AMXSY TD
3	THE UNIVERSITY OF TEXAS AT AUSTIN CENTER FOR ELECTROMECHANICS J PRICE A WALLS J KITZMILLER	1	APG MD 21005-5071 DIRECTOR US ARMY RESEARCH LAB AMSRL OP AP L APG MD 21005 5066
	10100 BURNET RD AUSTIN TX 78758-4497	115	DIR USARL AMSRL CI AMSRL CI H
3	VA POLYTECHNICAL INSTITUTE & STATE UNIVERSITY DEPT OF ESM M W HYER K REIFSNIDER R JONES BLACKSBURG VA 24061-0219		W STUREK AMSRL CI S A MARK AMSRL CS IO FI M ADAMSON AMSRL SL B J SMITH AMSRL SL BA
1	NORTH CAROLINA STATE UNIVERSITY CIVIL ENGINEERING DEPT W RASDORF PO BOX 7908 RALEIGH NC 27696-7908		AMSRL SL BL D BELY R HENRY AMSRL SL BG A YOUNG AMSRL SL I AMSRL WM B
1	UNIVERSITY OF MARYLAND DEPT OF AEROSPACE ENGINEERING ANTHONY J VIZZINI COLLEGE PARK MD 20742		A HORST E SCHMIDT AMSRL WM BA W D AMICO F BRANDON
1	DREXEL UNIVERSITY ALBERT S D WANG 32ND AND CHESTNUT STREETS PHILADELPHIA PA 19104		AMSRL WM BC P PLOSTINS D LYON J NEWILL S WILKERSON
1	SOUTHWEST RSCH INSTITUTE ENGR & MATL SCIENCES DIV J RIEGEL 6220 CULEBRA RD PO DRAWER 28510 SAN ANTONIO TX 78228-0510		A ZIELINSKI AMSRL WM BD B FORCH R FIFER R PESCE RODRIGUEZ B RICE

NO. OF

ORGANIZATION <u>COPIES</u>

ABERDEEN PROVING GROUND (CONT)

AMSRL WM MB

ABERDEEN PROVING GROUND (CONT)

ERIGAS

J SANDS

D SPAGNUOLO

W SPURGEON

J TZENG

E WETZEL

A ABRAHAMIAN

M BERMAN

A FRYDMAN

TLI

W MCINTOSH

E SZYMANSKI

AMRSL WM MC

J BEATTY

J SWAB

E CHIN

J MONTGOMERY

A WERESCZCAK

J LASALVIA

J WELLS

AMSRL WM MD

WROY

S WALSH

AMSRL WM T

B BURNS

AMSRL WM TA

W GILLICH

THAVEL

J RUNYEON

M BURKINS **E HORWATH**

B GOOCH

W BRUCHEY

AMSRL WM TC

R COATES

AMSRL WM TD

A DAS GUPTA

T HADUCH

T MOYNIHAN

FGREGORY A RAJENDRAN

M RAFTENBERG

M BOTELER

T WEERASOORIYA

D DANDEKAR

A DIETRICH

AMSRL WM BE

G WREN

C LEVERITT

D KOOKER

AMSRL WM BR

C SHOEMAKER

J BORNSTEIN

AMSRL WM M

D VIECHNICKI

G HAGNAUER

J MCCAULEY

B TANNER

AMSRL WM MA

R SHUFORD

P TOUCHET

N BECK TAN

DFLANAGAN

L GHIORSE

D HARRIS S MCKNIGHT

P MOY

S NGYUEN

PPATTERSON

G RODRIGUEZ

A TEETS

R YIN

AMSRL WM MB

B FINK

J BENDER

TBLANAS

T BOGETTI

R BOSSOLI L BURTON

K BOYD

S CORNELISON

P DEHMER

R DOOLEY

W DRYSDALE

G GAZONAS

S GHIORSE

D GRANVILLE

D HOPKINS

C HOPPEL

DHENRY

R KASTE

M KLUSEWITZ M LEADORE

R LIEB

NO. OF

COPIES ORGANIZATION

ABERDEEN PROVING GROUND (CONT)

AMSRL WM TE

A NIILER

J POWELL

AMSRL SS SD

H WALLACE

AMSRL SS SE R

R CHASE

AMSRL SS SE DS

R REYZER

R ATKINSON

AMSRL SE L

R WEINRAUB

J DESMOND

D WOODBURY

NO. OF COPIES ORGANIZATION

- 1 R MARTIN
 MERL
 LTD
 TAMWORTH RD
 HERTFORD SG13 7DG
- 1 PW LAY
 SMC SCOTLAND
 DERA ROSYTH
 ROSYTH ROYAL DOCKYARD
 DUNFERMLINE FIFE KY 11 2XR
 UK
- 1 T GOTTESMAN
 CIVIL AVIATION ADMINSTRATION
 PO BOX 8
 BEN GURION INTERNL AIRPORT
 LOD 70150 ISRAEL
- 1 S ANDRE
 AEROSPATIALE
 A BTE CC RTE MD132
 316 ROUTE DE BAYONNE
 TOULOUSE 31060
 FRANCE
- 1 J BAUER
 DAIMLER BENZ AEROSPACE
 D 81663 MUNCHEN
 MUNICH
 GERMANY
- 3 DRA FORT HALSTEAD
 PETER N JONES
 DAVID SCOTT
 MIKE HINTON
 SEVEN OAKS KENT TN 147BP
 UNITED KINGDOM
- 1 MR FRANCOIS LESAGE
 DEFENSE RESEARCH ESTAB
 VALCARTIER
 PO BOX 8800
 COURCELETTE QUEBEC COA
 IRO CANADA

NO. OF COPIES ORGANIZATION

- 2 ROYAL MILITARY COLLEGE OF SCIENCE SHRIVENHAM D BULMAN B LAWTON SWINDON WILTS SN6 8LA UNITED KINGDOM
- 1 SWISS FEDERAL ARMAMENTS
 WKS
 WALTER LANZ
 ALLMENDSTRASSE 86
 3602 THUN
 SWITZERLAND
- 1 PROFESSOR SOL BODNER
 ISRAEL INSTITUTE OF
 TECHNOLOGY
 FACULTY OF MECHANICAL ENGR
 HAIFA 3200 ISRAEL
- DSTO MATERIALS RSRCH LAB
 DR NORBERT BURMAN NAVAL
 PLATFORM VULNERABILITY SHIP
 STRUCTURES & MATERIALS DIV
 PO BOX 50
 ASCOT VALE VICTORIA
 AUSTRALIA 3032
- PROFESSOR EDWARD CELENS
 ECOLE ROYAL MILITAIRE
 AVE DE LA RENAISSANCE 30
 1040 BRUXELLE
 BELGIQUE
- 1 DEF RES ESTABLISHMENT
 VALCARTIER
 ALAIN DUPUIS
 2459 BOULEVARD PIE XI NORTH
 VALCARTIER QUEBEC
 CANADA
 PO BOX 8800 COURCELETTE
 GOA IRO QUEBEC CANADA

NO. OF COPIES ORGANIZATION

- 1 INSTITUT FRANCO ALLEMAND DE RECHERCHES DE SANIT LOUIS DE MARC GIRAUD RUE DU GENERAL CASSAGNOU BOITE POSTALE 34 F 68301 SAINT LOUIS CEDEX FRANCE
- J MANSON
 ECOLE POLYTECH
 DMX LTC
 CH 1015 LAUSANNE SWITZERLAND
- 1 TNO PRINS MAURITS LAB DR ROB IJSSELSTEIN LANGE KLEIWEG 137 PO BOX 45 2280 AA RUSWUK THE NETHERLANDS
- 2 FOA NAT L DEFENSE
 RESEARCH ESTAB
 DR BO JANZON
 R HOLMLIN
 DIR DEPT OF WEAPONS &
 PROTECTION
 S 172 90 STOCKHOLM
 SWEDEN
- 2 DEFENSE TECH & PROC AGENCY
 GRND
 MR I CREWTHER
 GENERAL HERZOG HAUS
 3602 THUN
 SWITZERLAND
- 1 MINISTRY OF DEFENCE
 RAFAEL
 DR MEIR MAYSELESS
 ARMAMENT DEVELOPMENT AUTH
 PO BOX 2250
 HAIFA 31021 ISRAEL
- 1 DR AKE PERSSON DYNAMEC RESEARCH AB PARADISGRND 7 S 151 36 SODERTALJE SWEDEN

NO. OF COPIES ORGANIZATION

- 1 ERNST MACH INSTITUT EMI DIRECTOR HAUPTSTRASSE 18 79576 WEIL AM RHEIN GERMANY
- 1 ERNST MACH INSTITUT EMI DR ALOIS STILP ECKERSTRASSE 4 7800 FREIBURG GERMANY
- DR IR HANS PASMAN
 TNO DEFENSE RESEARCH
 POSTBUS 6006
 2600 JA DELFT
 THE NETHERLANDS
- DR BITAN HIRSCH
 TACHKEMONY ST 6
 NETAMUA 42611
 ISRAEL
- 1 PROF DR MANFRED HELD
 DEUTSCHE AEROSPACE AG
 DYNAMICS SYSTEMS
 PO BOX 1340
 D 86523 SCHROBENHAUSEN
 GERMANY

Form Approved REPORT DOCUMENTATION PAGE OMB No. 0704-0188 Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, and completing and reviewing the collection of information. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing this burden, to Washington Headquarters Services, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway. Suite 1204. Aritination, VA 22202-4302, and to the Office of Management and Budget, Paperwork Reduction Project(0704-0188). Washington, DC 20503. 2. REPORT DATE 3. REPORT TYPE AND DATES COVERED 1. AGENCY USE ONLY (Leave blank) Final, January 1996 - December 1996 June 2000 5. FUNDING NUMBERS 4. TITLE AND SUBTITLE On the Influence of Moisture on Dielectric Properties of Polyetheretherketone (PEEK) Carbon-Fiber Composites AH42 6. AUTHOR(S) Bruce K. Fink, Roy L. McCullough,* and John W. Gillespie Jr.* 8. PERFORMING ORGANIZATION 7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) REPORT NUMBER U.S. Army Research Laboratory ARL-TR-2236 ATTN: AMSRL-WM-MB Aberdeen Proving Ground, MD 21005-5069 10.SPONSORING/MONITORING 9. SPONSORING/MONITORING AGENCY NAMES(S) AND ADDRESS(ES) AGENCY REPORT NUMBER 11. SUPPLEMENTARY NOTES *University of Delaware, Newark, DE 19716 12b. DISTRIBUTION CODE 12a. DISTRIBUTION/AVAILABILITY STATEMENT Approved for public release; distribution is unlimited. 13. ABSTRACT (Maximum 200 words) An analysis of local and global mechanisms of heat generation and distribution in carbon-fiber-based composites subjected to an alternating magnetic field has shown that heating is dependent upon the dielectric properties of the polymer matrix. These properties were investigated as functions of temperature, frequency, and moisture content. The results indicate little dependence of the dielectric constant on temperature or frequency, while the loss tangent exhibited a substantial dependence on both frequency and temperature. A substantial dependence of loss tangent on moisture content in polyetheretherketone (PEEK) was found. 15. NUMBER OF PAGES 14. SUBJECT TERMS 16. PRICE CODE carbon, composites, dielectric PEEK, moisture induction 19. SECURITY CLASSIFICATION 20. LIMITATION OF ABSTRACT 18. SECURITY CLASSIFICATION 17. SECURITY CLASSIFICATION

UNCLASSIFIED

OF REPORT

OF THIS PAGE

UNCLASSIFIED

OF ABSTRACT

UNCLASSIFIED

INTENTIONALLY LEFT BLANK.

USER EVALUATION SHEET/CHANGE OF ADDRESS

This Laboratory undertakes a continuing effort to improve the quality of the reports it publishes. Your comments/answers to the items/questions below will aid us in our efforts.

1. ARL Report Num	ber/Author ARL-TR-2236 (Fink)	Date of Report June 2000
2. Date Report Rece	ived	
	tisfy a need? (Comment on purpose, relate	d project, or other area of interest for which the report will be
4. Specifically, how	is the report being used? (Information sour	ce, design data, procedure, source of ideas, etc.)
	es achieved, etc? If so, please elaborate	vings as far as man-hours or dollars saved, operating costs
6. General Comment	s. What do you think should be changed to	improve future reports? (Indicate changes to organization,
	Organization	
CURRENT	Name	E-mail Name
ADDRESS	Street or P.O. Box No.	
	City, State, Zip Code	
7. If indicating a Cha or Incorrect address b	•	se provide the Current or Correct address above and the Old
	Organization	
OLD	Name	
ADDRESS	Street or P.O. Box No.	
	City, State, Zip Code	
	(Pomova this shoot, fold as indic	oted tape closed and mail)

(Remove this sheet, fold as indicated, tape closed, and mail.)
(DO NOT STAPLE)

DEPARTMENT OF THE ARMY

OFFICIAL BUSINESS

BUSINESS REPLY MAIL FIRST CLASS PERMIT NO 0001, APG, MD

POSTAGE WILL BE PAID BY ADDRESSEE

DIRECTOR
US ARMY RESEARCH LABORATORY
ATTN AMSRL WM MB
ABERDEEN PROVING GROUND MD 21005-5069

NO POSTAGE
NECESSARY
IF MAILED
IN THE
UNITED STATES